

ET662465014US

INTEGRATED ELECTRO-
LUMINESCENT BIOCHIP

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1 This is a continuation-in-part of an application filed
2 July 8, 1998 under Serial No. 08/112,398, which is a
3 continuation-in-part of an application filed November 17,
4 1995 under Serial No. 08/560,380, which is a divisional
5 application of a patent application filed June 30, 1993
6 under Serial No. 08/084,876.

7 BACKGROUND OF THE INVENTION

8 The invention relates to the field of detectors for
9 analysis of biological samples located on biochips.

10 U. S. Patent No. 5,770,029 teaches an integrated
11 electro-phoretic micro-devices each of which includes at
12 least an enrichment channel and a main electro-phoretic
13 flow-path are provided. In the subject integrated devices,
14 the enrichment channel and the main electro-phoretic flow-
15 path are positioned so that waste fluid flows away from said

1 main electro-phoretic flow-path through a discharge outlet.

2 The subject devices find use in a variety of electro-
3 phoretic applications, including clinical assays.

4 U. S. Patent No. 6,159,681 teaches compositions and
5 methods which are provided for performing regional analysis
6 of biologic materials. The methods provided herein employ a
7 photo-resist layer that is established over a biologic
8 material (which may be immobilized on a substrate). Regions
9 of interest are selected and irradiated to expose specific
10 regions of biologic material. Exposed biologic material may
11 then be selectively analyzed using any of a variety of
12 analytic methods.

13 U. S. Patent 6,160,618 teaches an apparatus for
14 analyzing samples on a slide which includes a slide mover
15 positioned to hold a slide, a imaging spectrometer
16 positioned in the path of light from the slide to split the
17 light line into a light array, a light amplifier may be
18 positioned between the imaging spectrometer and a camera, is

1 disclosed. The camera can detect the entire spectrum of
2 light produced by the imaging spectrometer.

3 U. S. Patent No. 6,110,676 teaches methods which are
4 qsuitable for detection, analysis and quantitation of
5 nucleic acid target sequences using probe based
6 hybridization assays and more specifically for suppressing
7 the binding of detectable nucleic acid probes or detectable
8 PNA probes to non-target nucleic acid sequences in an assay
9 for a target nucleic acid sequence to thereby improve the
10 reliability, sensitivity and specificity of the assay. The
11 methods, kits and compositions of this invention are
12 particularly well suited to the detection and analysis of
13 nucleic acid point mutations.

14 U. S. Patent No. 6,245,507 teaches a hyper-spectral
15 imaging apparatus. The apparatus employs an apparatus for
16 multi-dye/base detection of a nucleic acid molecule coupled
17 to a solid surface.

18 U. S. Patent No. 6,245,506 teaches the use of the

1 discovery that the sequence of monomers in a polymeric
2 biomolecule can be determined in a self-contained, high
3 pressure reaction and detection apparatus, without the need
4 for fluid flow into or out from the apparatus. The pressure
5 is used to control the activity of enzymes that digest the
6 polymeric biomolecule to yield the individual monomers in
7 the sequence in which they existed in the polymer. High
8 pressures modulate enzyme kinetics by reversibly inhibiting
9 those enzymatic processes. These processes result in a
10 higher average activation volume, when compared to the
11 ground state, and reversibly accelerating those processes
12 which have lower activation volumes than the ground state.
13 Modulating the pressure allows the experimenter to precisely
14 control the activity of the enzyme. Conditions can be found,
15 for example, where the enzyme removes only one monomer
16 (e.g., a nucleotide or amino acid) from the biomolecule
17 before the pressure is again raised to a prohibitive level.
18 The identity of the single released nucleotide or amino acid

1 can be determined using a detector that is in communication
2 with a probe in the detection zone within the reaction
3 vessel.

4 U. S. Patent No. 6,240,790 teaches a microanalysis
5 device. The device has a plurality of sample processing
6 compartments is described for use in liquid phase analysis.
7 A microanalysis device system, comprising a plurality of
8 interconnected microanalysis devices. The device is formed
9 by microfabrication of microstructures in novel support
10 substrates.

11 Detection devices that detect and locate samples
12 contained on a biochip via laser light sources and laser
13 scanners are well known in the art. These detection devices
14 require the samples to be labeled by a fluorescent tag.
15 Typically, these detection devices rely on laser light
16 sources to excite the samples that are labeled by a
17 fluorescent tag and causes biologically active samples to
18 output emitted light waves. The laser source is scanned to

1 serially excite each sample on the biochip to detect any
 2 emitted light waves from the samples that are biologically
 3 active. Unfortunately, these detection devices utilizing
 4 either the laser light source or the laser scanner suffers
 5 from various drawbacks. First, laser scanners utilized to
 6 detect the emitted light waves from the exited samples on
 7 the biochip typically require wait times upwards of five
 8 minutes for sufficient resolution. Because laser scanners
 9 operate as a serial scanning device by sequentially
 10 detecting one sample at a time on the surface of the
 11 biochip, laser scanners are inherently inefficient at
 12 detecting the emitted light waves from an array of samples.

13 Further, laser light sources utilized within the
 14 detection devices inherently only emit coherent light-waves.
 15 The light-waves span over an extremely narrow range of
 16 wavelengths. Fluorescent tags are generally responsive to a
 17 single frequency of light or light from a narrow frequency
 18 band. Thus, the use of the laser light sources severely

1 limits the flexibility of those detection devices because
2 only one type of fluorescent tag can be used. In order to
3 use other tags additional laser sources must be used. In
4 order to evaluate a biochip that has been treated with
5 multiple tags, a long duration scan cycle must be performed
6 for each one of the required laser sources. If samples on a
7 biochip were labeled with two different fluorescent tags and
8 the different tags required light waves with substantially
9 different excitation wavelengths, analyzing these samples
10 would require the user to change laser light sources the
11 analysis of all the samples were completed. Additionally,
12 to be able to handle samples labeled with different
13 fluorescent tags with differing excitation wavelengths, the
14 user is required to have access to a variety of laser light
15 sources. Since laser light sources are costly and
16 specialized items, there are substantial costs and
17 inconveniences associated with utilizing these prior
18 detection devices.

1 Therefore, it is desirable to have an ability to detect
2 and locate samples labeled with multiple tags contained on a
3 biochip, without the need for a laser light source. It is
4 also desirable have an ability to detect and locate samples
5 labeled with a tag contained on a biochip, without the need
6 for a serial scanning device.

7 U. S. Patent No. 6,197,503 teaches a self-contained
8 miniature DNA biosensor. The biosensor detects specific
9 molecular targets, particularly suitable for detection of
10 nucleic acids. Hybridized DNA may be detected without
11 external monitoring or signal transmission. The biosensor
12 is a biochip and includes multiple biological sensing
13 elements such as DNA probes, excitation micro-lasers, a
14 sampling wave-guide equipped with optical detectors
15 (fluorescence and Raman), integrated electro-optics, and a
16 bio-telemetric radio frequency signal generator. The novel
17 integrated circuit biochip micro-system (ICBM) is suitable
18 for gene analysis and will allow rapid, large-scale and

1 cost-effective production of gene biochips.

2 U. S. Patent No. 6,280,946 teaches PNA probes. The
3 probes pertain to the universal detection of bacteria and/or
4 eucarya. Preferred universal probes for the detection of
5 bacteria comprise a probing nucleo-base sequence selected
6 from the group consisting of CTG-CCT-CCC-GTA-GGA; TAC-CAG-
7 GGT-ATC-TAA-T; CAC-GAG-CTG-ACG-ACA and CCG-ACA-AGG-AAT-TTC.

8 Preferred universal probes for the detection of eucarya
9 include a probing nucleo-base sequence selected from the
10 group consisting of ACC-AGA-CTT-GCC-CTC-C; GGG-CAT-CAC-AGA-
11 CCT-G; TAG-AAA-GGG-CAG-GGA and TAC-AAA-GGG-CAG-GGA. The PNA
12 probes, probe sets, methods and kits of this invention are
13 particularly well suited for use in multiplex PNA-FISH
14 assays wherein the bacteria and/or eucarya in a sample can
15 be individually detected, identified or quantitated. Using
16 exemplary assays described herein, the total number of
17 colony forming units (CFU) of bacteria and/or eucarya can be
18 rapidly determined.

1 U. S. Patent No. 6,238,624 teaches a self-addressable,
2 self-assembling microelectronic device. The device is
3 designed and fabricated to actively carry out and control
4 multi-step and multiplex molecular biological reactions in
5 microscopic formats. These reactions include nucleic acid
6 hybridizations, antibody/antigen reactions, diagnostics, and
7 biopolymer synthesis. The device can be fabricated using
8 both micro-lithographic and micro-machining techniques. The
9 device can electronically control the transport and
10 attachment of specific binding entities to specific micro-
11 locations. The specific binding entities include molecular
12 biological molecules such as nucleic acids and polypeptides.
13 The device can subsequently control the transport and
14 reaction of analytes or reactants at the addressed specific
15 micro-locations. The device is able to concentrate analytes
16 and reactants, remove non-specifically bound molecules,
17 provide stringency control for DNA hybridization reactions,
18 and improve the detection of analytes. The device can be

1 electronically replicated.

2 U. S. Patent No. 6,271,042 teaches a biochip detection
3 system. The biochip detection system detects and locates
4 samples that are labeled with multiple fluorescent tags and
5 are located on a biochip. This biochip detection system
6 includes a charge coupled device (CCD) sensor, a broad-
7 spectrum light source, a lens, a light source filter, and a
8 sensor filter. The CCD sensor includes two-dimensional CCD
9 arrays to simultaneously detect light waves from at least a
10 substantial portion of the biochip. The broad-spectrum
11 light source is optically coupled to the CCD sensor and is
12 configured to be utilized with a variety of different
13 fluorescent tags. The tags have differing excitation
14 wavelengths.

15 U. S. Patent No. 4,983,369 a process for producing
16 highly uniform microspheres of silica having an average
17 diameter of 0.1-10 microns from the hydrolysis of a silica
18 precursor, such as tetraalkoxysilanes, which is

1 characterized by employing precursor solutions and feed
2 rates which initially yield a two-phase reaction mixture.

3 U. S. Patent No. 4,943,425 teaches a method of making
4 high purity, dense silica of large particles size.

5 Tetraethylorthosilicate is mixed with ethanol and is added
6 to a dilute acid solution having a pH of about 2.25. The
7 resulting solution is digested for about 5 hours, then 2N
8 ammonium hydroxide is added to form a gel at a pH of 8.5.
9 The gel is screened through an 18-20 mesh screen, vacuum
10 baked, calcined in an oxygen atmosphere and finally heated
11 to about 1200 C in air to form a large particle size, high
12 purity, dense silica.

13 U. S. Patent No. 4,965,x91 teaches a sol-gel procedure
14 is described for making display devices with luminescent
15 films. The procedure typically involves hydrolysis and
16 polymerization of an organo-metallic compound together with
17 selected luminescent ions, and coating of a substrate and
18 then heat treatment to form a polycrystalline layer.

1 U. S. Patent No. 4,931,312 teaches luminescent thin
2 films which are produced by a sol-gel process in which a
3 gellable liquid is applied to a substrate to form a thin
4 film; gelled and heated to remove volatile constituents and
5 form a polycrystalline luminescent material.

6 U. S. Patent No: 4,997,286 teaches an apparatus for
7 measuring temperature in a region of high temperature which
8 includes a sensor made from a fluorescent material, located
9 within the region of high temperature. The fluorescent decay
10 time of the fluorescent material is dependent upon the
11 temperature of the fluorescent material.

12 U. S. Patent No. 4,948,214 teaches an array of
13 individual light emitters of a LED linear array each of
14 which is imaged by a discrete step-index light guide and
15 gradient index micro-lens device. The light guides consist
16 of high refractive index cores each surrounded by low
17 refractive index matter. A multiplicity of light guides are
18 deposited in channels formed in a host material, such as a

1 silicon wafer. The host material between adjacent channels
2 functions as an opaque separator to prevent cross-talk
3 between adjacent light guides.

4 U. S. Patent No. 4,925,275 teaches a liquid crystal
5 color display which provides a transmitted light output that
6 is of one or more-colors, black, and/or white, as a-function
7 of the color of the incident light input and controlled
8 energization or not of respective optically serially
9 positioned liquid crystal color layers and/or -multicolor
10 composite liquid crystal color layer(s) in the display. In
11 one case the display includes a plurality of liquid crystal
12 color layers each being dyed a different respective color,
13 and apparatus for selectively applying a prescribed input,
14 such as an electric field, to a respective layer or layers
15 or to a portion or portions thereof. Each liquid crystal
16 layer includes plural volumes of operationally nematic
17 liquid crystal material in a containment medium that tends
18 to distort the natural liquid crystal structure in the

1 absence of a prescribed input, such as an electric field,
2 and pleochroic dye is included or mixed with the liquid
3 crystal material in each layer. Each layer is differently
4 colored by the dye so as to have a particular coloring
5 effect on light incident thereon. Exemplary layer colors
6 may be yellow, cyan and magenta.

7 U. S. Patent No. 4,957,349 teaches an active matrix
8 screen for the color display of television images or
9 pictures, control system which utilizes the electrically
10 controlled birefringence effect and includes an assembly
11 having a nematic liquid crystal layer with a positive
12 optical anisotropy between an active matrix having
13 transparent control electrodes and a transparent counter
14 electrode equipped with colored filters and two polarizing
15 means, which are complimentary of one another and are
16 located on either side of the assembly.

17 U. S. Patent No. 4,948,843 teaches dye-containing
18 polymers in which the dyes are organic in nature are

1 incorporated into glasses produced by a sol-gel technique.
2 The glasses may be inorganic or organic-modified metal oxide
3 heteropolycondensates. The dye-containing polymers are
4 covalently bonded to the glass through a linking group.
5 These products can be used to make optically clear colored
6 films which can be employed in the imaging, optical, solar
7 heat energy and related arts.

8 U. S. Patent No. 5,598,058 teaches a thick-film multi-
9 color electroluminescent display which includes a
10 transparent substrate, a transparent electrode deposited on
11 the substrate, a phosphor layer deposited on the transparent
12 electrode having two regions having different compositions
13 providing visually distinct spectra of light when placed in
14 a common electric field, a dielectric layer deposited on the
15 phosphor layer, and a second electrode deposited on the
16 dielectric layer. The phosphor layer may be composed of a
17 marbled-ink having a mixture of a first phosphor ink and a
18 second phosphor ink having different compositions providing

1 visually distinct spectra of light when placed in a common
 2 electric field. The phosphor layer may be composed of at
 3 least two halftone screen prints corresponding to at least
 4 two phosphor compositions providing visually distinct
 5 spectra of light when placed in a common electric field.

6 U. S. Patent No. 5,602,445 teaches a bright, short
 7 wavelength blue-violet phosphor for electroluminescent
 8 displays which includes an alkaline-based halide as a host
 9 material and a rare earth as a dopant. The host alkaline
 10 chloride can be chosen from the group II alkaline elements,
 11 particularly strontium chloride (SrCl_2) or calcium
 12 chloride (CaCl_2), which, with a europium (Eu) or cerium
 13 (Ce) rare earth dopant, electroluminesces at a peak
 14 wavelength of 404 and 367 nanometers (nm) respectively. The
 15 resulting emissions have CIE chromaticity coordinates which
 16 lie at the boundary of the visible range for the human eye
 17 thereby allowing a greater range of colors for full color
 18 flat panel electroluminescent (FPEL) displays.

1 U. S. Patent No. 5,719,467 teaches an organic
2 electroluminescent device which has a conducting polymer
3 layer beneath the hole-transport layer. A conducting
4 polymer layer of doped polyaniline (PANI) is spin-cast onto
5 an indium-tin oxide (ITO) anode coating on a glass
6 substrate. Then a hole-transport layer, for example TPD or
7 another aromatic tertiary amine, is vapor-deposited onto the
8 conducting polymer layer, followed by an electron transport
9 layer and a cathode. Polyester may be blended into the PANI
10 before spin-casting and then removed by a selective solvent
11 after the spincasting leaving a microporous layer of PANI
12 on the anode. The conducting polymer layer may instead be
13 made of a pi-conjugated oxidized polymer or of TPD dispersed
14 in a polymer binder that is doped with an electron-
15 withdrawing compound. An additional layer of copper-
16 phthalocyanine, or of TPD in a polymer binder may be
17 disposed between the conducting polymer layer and the hole
18 transport layer. The conducting polymer layer may serve as

1 the anode, in which case the ITO is omitted.

2 U. S. Patent No. 5,717,289 teaches a thin film
3 electroluminescent element which has a color changing layer
4 doped with green luminescent material and red fluorescent
5 material and separated from an electroluminescent layer for
6 generating blue light for converting the blue light to green
7 light and the green light to red light, and the separation
8 results in reduction of trapping center in the electro-
9 luminescent layer.

10 U. S. Patent No. 5,711,898 teaches a blue-green
11 emitting ZnS: Cu, Cl phosphor which is made by doping the
12 phosphor with small amounts of gold and increasing the
13 amount of low intensity milling between firing steps. The
14 phosphor has better half-life and brightness characteristics
15 while maintaining its desired emission color.

16 U. S. Patent No. 5,705,888 teaches an electro-
17 luminescent device which is composed of polymeric LEDs
18 having an active layer of a conjugated polymer and a

1 transparent pelymeric electrode layer having electro-
2 conductive areas as electrodes. Like the active layer, the
3 electrode layer can be manufactured in a simple manner by
4 spin coating. The electrode layer is structured into
5 conductive electrodes by exposure to UV light. The
6 electrodes jointly form a matrix of LEDs for a display. When
7 a flexible substrate is used, a very bendable EL device is
8 obtained.

9 U. S. Patent No. 5,705,285 teaches an organic electro-
10 luminescent display device which includes a plurality of
11 pixels including a substrate upon which is disposed on a
12 plurality of different light influencing elements.
13 Deposited atop each light-influencing element is an organic
14 electroluminescent display element which is adapted to emit
15 light of a preselected wavelength. A layer of an
16 insulating, planarizing material may optionally be disposed
17 between the light influencing elements and the OED. Each
18 light-influencing element generates a different effect in

1 response to light of a preselected incident thereon. In
2 this way, it is possible to achieve a red, green, blue
3 organic electroluminescent display assembly using a single
4 organic electroluminescent display device.

5 U. S. Patent No. 5,705,284 teaches a thin film
6 electroluminescence device which is characterized in that as
7 a light emitting layer material or charge 'injection layer
8 material, a polymer film having at least one of a light
9 emitting layer function, a charge transport function and a
10 charge injection function, and having a film thickness of
11 not: more than 0.5 micon is prepared by the vacuum
12 evaporation method and used.

13 U. S. Patent No. 5,703,436 teaches a multicolor organic
14 light-emitting device. The LED device employs vertically
15 stacked layers of double hetero-structure devices which are
16 fabricated from organic compounds. The vertical stacked
17 structure is formed on a glass base having a transparent
18 coating of ITO or similar metal to provide a substrate.

1 Deposited on the substrate is the vertical stacked
2 arrangement of three double hetero-structure devices, each
3 fabricated from a suitable organic material. Stacking is
4 implemented such that the double hetero-structure with the
5 longest wavelength is on the top of the stack. This
6 constitutes the device emitting red light on the top with
7 the device having the shortest wavelength, namely, the
8 device emitting blue light, on the bottom of the stack.
9 Located between the red and blue device structures is the
10 green device structure. The devices are configured as
11 stacked to provide a staircase profile whereby each device
12 is separated from the other by a thin transparent conductive
13 contact layer to enable light emanating from each of the
14 devices to pass through the semitransparent contacts and
15 through the lower device structures while further enabling
16 each of the devices to receive a selective bias. The devices
17 are substantially transparent when de-energized, making them
18 useful for heads-up display applications.

1 U. S. Patent No. 5,702,643 teaches a ZnS:Cu
 2 electroluminescent phosphor which has a halflife of at least
 3 about 900 hours. The half-life improvement is made by
 4 doping the phosphor with minor amounts of gold and
 5 substantially increasing the amount of low intensity milling
 6 between firing steps. The phosphor has a dramatically
 7 longer halflife without sacrificing brightness or exhibiting
 8 large shifts in emission color.

9 U. S. Patent No. 5,700,592 teaches an electro-
 10 luminescent edge emitting device which has an improved
 11 operational life and electroluminescent efficiency includes
 12 a host material composed of at least two Group II elements
 13 and at least one element selected from Group VIA. The host
 14 material is doped with at least one of the rare earth
 15 elements in its 3+ or 2+ oxidation state. Two Group IIB
 16 elements may be selected, namely cadmium and zinc. Three
 17 Group IIA elements, magnesium, calcium and strontium, may
 18 be selected as the host material. The Group VIA element is

1 sulfide and/or selenide. The dopant is composed of one, two
2 or three elements selected from the rare earth elements
3 (lanthanides). The dopants may include Mn.sup.2+ and one or
4 two of the lanthanides.

5 U. S. Patent No. 5,700,591 teaches a phosphor thin film
6 of a compound of zinc, cadmium, manganese or alkaline earth
7 metals and an element of group VI which is sandwiched by
8 barrier layers having a larger energy gap than that of the
9 phosphor thin film, and a plurality of the sandwich
10 structures are accumulated thicknesswise to constitute a
11 light-emitting device. The phosphor thin film ensures the
12 confinement of injected electrons and holes within the
13 phosphor thin film. The light-emitting device has a high
14 brightness and a high efficiency.

15 U. S. Patent No. 5,693,962 teaches an organic full
16 color light emitting diode array which includes a plurality
17 of spaced apart, light transmissive electrodes formed on a
18 substrate, a plurality of cavities defined on top of the

1 electrodes and three electroluminescent media designed to
2 emit three different hues deposited in the cavities. A
3 plurality of spaced metallic electrodes arranged orthogonal
4 to the transmissive electrodes and formed to seal each of
5 the cavities, thereby, sealing the electroluminescent media
6 in the cavities, with a light transmissive anodic electrode
7 at the bottom of each cavity and an ambient stable cathodic
8 metallic electrode on the top of each cavity.

9 U. S. Patent No. 5,683,823 teaches an electro-
10 luminescent device. The device includes an anode, a
11 positive-hole transporting layer made of an organic
12 compound, a fluorescent-emitting layer made of an organic
13 compound and a cathode. The fluorescent emitting layer
14 includes a red light-emitting material uniformly dispersed
15 in a host emitting material. The host emitting material is
16 adapted to emit in the blue green regions so that the light
17 produced by this device is substantially white.

18 U. S. Patent No. 5,677,594 teaches an electro-

1 luminescent phosphor which is sandwiched by a pair of
2 insulating layers which are sandwiched by a pair of
3 electrode layers to provide an AC TFEL device. The phosphor
4 consists of a host material and an activator dopant that is
5 preferably a rare earth. The host material is an alkaline
6 earth sulfide, an alkaline earth selenide or an alkaline
7 earth sulfide selenide that includes a Group 3A metal
8 selected from aluminum, gallium and indium. The phosphor is
9 preferably fabricated by first depositing a layer of the
10 alkaline earth sulfide, alkaline earth selenide or alkaline
11 earth sulfide selenide including the rare earth dopant
12 therein, depositing thereon an overlayer selected from an
13 alkaline earth thiogallate, an alkaline earth thioindate, an
14 alkaline earth thioaluminate, an alkaline earth
15 selenoaluminate, an alkaline earth selenoindate, or an
16 alkaline earth selenogallate. The two layers are annealed
17 at a temperature preferably between 750 and 850 degrees C.

18 U. S. Patent No. 5,675,217 teaches a color EL device

1 which includes a substrate, a first electrode formed on the
2 substrate, a first insulating layer formed on the first
3 electrode, a phosphorous layer formed on the first
4 insulating layer and having inserted therein one or more
5 intermediate insulating layers, a second insulating layer
6 formed on the phosphorous layer and a second electrode
7 formed on the second insulating layer.

8 U. S. Patent No. 5,672,937 teaches flexible translucent
9 electro-conductive plastic film electrodes which are
10 produced by perforating a normally nonconductive translucent
11 plastic film, and then applying to both surfaces of the film
12 thin layers of a conductive metal oxide such as indium-tin
13 oxide. The conductive layers communicate through the
14 perforations to form an electro-conductive film electrode
15 useful with an electro-luminescent layer and a rear
16 electrode to form lights, signs and similar electro-
17 luminescent laminates.

18 U. S. Patent No. 5,670,839 teaches UV light of

1 increased luminous intensity. Layered on one surface of a
2 translucent substrate are a transparent electrode, a first
3 insulating layer, an EL layer, a second insulating layer,
4 and a metal electrode, in that order. A compound of the
5 general formula: $\text{Zn}_{1-x}\text{Mg}_x\text{S}$ is selected as a
6 host material of the EL layer, and Gd or a Gd compound is
7 selected as the luminescence center. The composition ratio x
8 of the compound selected as a host material is selected to
9 be within the range of $0.33 \leq x < 1$, and preferably
10 within the range of from 0.4-0.8, inclusive. This selection
11 allows the band gap energy of the host material to be higher
12 than the band gap energy of the luminescence center, thus
13 preventing the absorption of the emitted light by the host
14 material and providing UV light of increased luminous
15 intensity.

16 U. S. Patent No. 5,667,905 teaches a solid-state
17 electro-luminescent device. The device includes a mixed
18 material layer formed of a mixture of silicon and silicon

1 oxide doped with rare earth ions so as to show intense room-
 2 temperature photo- and electro-luminescence. The
 3 luminescence is due to internal transitions of the rare
 4 earth ions. The mixed material layer has an oxygen content
 5 ranging from 1 to 65 atomic percent and is produced by vapor
 6 deposition and rare earth ions implant. A separated implant
 7 with elements of the V or III column of the periodic table
 8 of elements gives rise to a PN junction. The so obtained
 9 structure is then subjected to thermal treatment in the
 10 range 400 to 1100 degrees C.

11 U. S. Patent No. 5,663,573 teaches light-emitting
 12 bipolar devices. The devices consist of a light-emitter
 13 formed from an electro-luminescent organic light-emitting
 14 material in contact with an insulating material. The light
 15 emitter is in contact with two electrodes that are
 16 maintained in spaced apart relation with each other. The
 17 light emitter can be formed as an integral mixture of light
 18 emitting materials and insulating materials or as separate

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1 layers of light-emitting and insulating materials. The
2 devices operate with AC voltage of less than twenty-four
3 volts and in some instances at less than five volts. Under
4 AC driving, the devices produce modulated light output which
5 can be frequency or amplitude modulated. Under DC driving,
6 the devices operate in both forward and reverse bias.

7 U. S. Patent No. 5,656,888 teaches a novel thin-film
8 electro-luminescent (TFEL) structure for emitting light in
9 response to the application of an electric field which
10 includes first and second electrode layers sandwiching a
11 TFEL stack, the stack including first and second insulator
12 layers and a phosphor layer that includes an alkaline earth
13 thiogallate doped with oxygen.

14 U. S. Patent No. 5,652,067 teaches an organic electro-
15 luminescent device which includes a substrate and formed
16 thereon a multi-layered structure successively having at
17 least an anode layer, an organic electro-luminescent layer
18 and a cathode layer, a sealing layer having at least one

1 compound selected from the group consisting of a metal
2 oxide, a metal fluoride and a metal sulfide is further
3 provided on the electrode layer formed later. A hole
4 injecting and transporting layer is preferably provided
5 between the anode layer and the organic electro-luminescent
6 layer. An electron injecting and transporting layer may
7 also be provided between the organic electro-luminescent
8 layer and the cathode layer. At least one layer of the hole
9 injecting and transporting layer, organic electro-
10 luminescent layer and electron injecting and transporting
11 layer may be formed of a poly-phosphazene compound or a
12 polyether compound or a polyphosphate compound having an
13 aromatic tertiary amine group in its main chain.

U. S. Patent No. 5,650,692 teaches an electro-luminescent device. The device includes a substrate and an electro-luminescent stack. The stack forms a step relative to the substrate. A transparent layer of protective material is placed atop the stack to bridge the step and

1 create a smooth edge profile along the edge. A
2 metallization layer is situated atop the layer of protective
3 material and is coupled to the electro-luminescent stack
4 through vias in the protective material.

5 U. S. Patent No. 5,648,181 teaches an inorganic thin
6 film EL device which includes on an insulating substrate, a
7 back electrode, an insulating layer, a light emission layer,
8 an insulating layer, and a transparent electrode formed on
9 the substrate in this order. The emission layer includes
10 lanthanum fluoride and at least one member selected from the
11 group consisting of rare earth element metals and compounds
12 thereof. The rare earth element is, for example, cerium,
13 praseodymium, neodium, samarium, europium, gadolinium,
14 terbium, dysprosium, holmium, erbium, thulium, ytterbium and
15 mixture thereof. The compounds maybe those compounds of the
16 rare earth elements and fluorine, chlorine, bromine, iodine
17 and oxygen. The rare earth element is preferably present in
18 the emission layer in an amount of from 5 to 90 wt

1 U. S. Patent No. 5,646,480 teaches an electro-
2 luminescent display panel which has a plurality of parallel
3 metal assist structures deposited on a glass substrate, a
4 plurality of parallel transparent electrodes are deposited
5 over and aligned with the metal assist structures such that
6 each metal assist structure is surrounded by a transparent
7 electrode. A conventional stack of dielectric and phosphor
8 layers and a plurality of metal electrodes is deposited
9 thereon to complete the electro-luminescent display panel.

10 U. S. Patent No. 5,645,948 teaches an organic EL device
11 which includes an anode and a cathode, and at least one
12 organic luminescent medium containing a compound of
13 benzazoles of the formula: ##STR1## wherein: n is an integer
14 of from 3 to 8; Z is O, NR or S; and R and R' are
15 individually hydrogen; alkyl of from 1 to 24 carbon atoms,
16 for example, propyl, t-butyl, heptyl, and the like; aryl or
17 hetero-atom substituted aryl of from 5 to 20 carbon atoms
18 for example, phenyl and naphthyl, furyl, thienyl, pyridyl,

1 quinolinyl and other heterocyclic systems; or halo such as
2 chloro, fluoro; or atoms necessary to complete a fused
3 aromatic ring; B is a linkage unit consisting of alkyl,
4 aryl, substituted alkyl, or substituted aryl which
5 conjugately or unconjugately connects the multiple
6 benzazoles together.

7 U. S. Patent No. 5,644,327 teaches an electro-
8 luminescent display formed on a ceramic substrate. The
9 substrate has a front ceramic surface and a back ceramic
10 surface. The ceramic substrate includes a metal core that
11 provides structural support, electrical ground, and heat
12 dissipation. Electro-luminescent cells are mounted on the
13 front ceramic surface and driver circuits for driving the
14 electro-luminescent cells are mounted on the back ceramic
15 surface. The driver circuits are positioned directly behind
16 the electro-luminescent cells. Connectors extend through
17 the ceramic substrate and the electro-luminescent cells to
18 different driver circuits. By positioning the driver

1 circuits close to the EL cells, the drive lines from the
2 drivers to the EL cells are short which allows for high
3 refresh rates and low resistance losses. Each of the driver
4 circuits can drive one electro-luminescent cell or a group
5 of electro-luminescent cells. EL display cells coupled to a
6 ceramic electrode can also be driven by a field emission
7 device or a low power electron beam.

8 U. S. Patent No. 5,643,829 teaches a multi-layer
9 electro-luminescence device which is formed by the steps of
10 forming a lower electrode with a predetermined pattern on a
11 substrate, forming a first insulation layer on the lower
12 electrode atop the substrate; forming a multiply luminescent
13 layer consisting of CaS and SrS on the first insulation
14 layer at the same temperature with that for the first
15 insulation layer; forming a second insulation film on the
16 luminescent layer-; and forming-an upper electrode with a
17 predetermined on the second insulation layer. In the
18 multiply luminescent layer, a plurality of CaS plies and a

1 plurality of SrS plies are formed in such a way that the CaS
2 plies and the SrS plies alternate with each other and the
3 outmost upper and lower plies are formed of CaS. The
4 constituent substances for the multiply luminescent layer,
5 CaS and SrS, can be deposited at the same temperature and
6 have similar lattice constants which can lead to a matched
7 interface between the CaS and SrS plies. By virtue of these
8 advantages, stresses imposed on the interface, including
9 thermal stress, can be significantly reduced. In addition,
10 the matched interface makes electrons be accelerated with
11 large energy, so that the fabricated multi-layer
12 luminescence device may show good quality.

13 U. S. Patent No. 5,643,685 teaches an electro-
14 luminescence element composed of a substrate, a first
15 electrode, a first insulating layer, a light-emitting layer,
16 a second insulating layer, and a second electrode in this
17 order and a process for producing the same are disclosed, in
18 which the light-emitting layer which includes a chemically

1 stable oxide material containing a plurality of elements,
2 the composition ratio of the elements constituting the oxide
3 material being substantially equal to that of the elements
4 charged, the light-emitting layer is formed by coating a
5 first insulating layer with a sol solution containing a
6 plurality of metal elements at a prescribed composition
7 ratio and heating the coating layer to form an oxide layer.

8 U. S. Patent No. 5,643,496 teaches an electro-
9 luminescent phosphor composed of copper activated zinc
10 sulfide having an average particle size less than 23
11 micrometers and a half-life equal to or greater than the
12 half-life of a second phosphor having a similar composition
13 and an average particle size of at least 25 micrometers.

14 U. S. Patent No. 5,641,582 teaches a thin-film EL
15 element which does not permit the color of the emitted light
16 to change irrespective of a change in the voltage, which
17 remains chemically stable and which emits light of high
18 brightness even on a low voltage. The element includes two

1 or more poly-crystalline thin light emitting layers and one
2 or more thin insulating layers. The interface between a thin
3 film and a thin film constituting a light emitting layer is
4 formed by epitaxial growth, and the electrical
5 characteristics of the element are equivalent to those of a
6 single circuit which includes two Zener diodes connected in
7 series, a capacitor connected in parallel with the serially
8 connected Zener diodes, and a capacitor connected to one end
9 of the capacitor.

10 U. S. Patent No. 5,635,308 teaches phenyl-anthracene
11 derivatives of the formula: A.sub.1 --L--A.sub.2 wherein
12 A.sub.1 and A.sub.2 each are a monophenylanthryl or
13 diphenylanthryl group and L is a valence bond or a divalent
14 linkage group, typically arylene are novel opto-electronic
15 functional materials. They are used as an organic compound
16 layer of organic EL device, especially a light emitting
17 layer for blue light emission.

18 U. S. Patent No. 5,635,307 teaches a thin-film EL

1 element having as a laminated luminescent composite a
2 configuration which includes at least a first layer and a
3 second layer wherein the first layer includes a compound
4 having a lattice constant, before lamination, larger than
5 that of a compound constituting the second layer, and
6 contains manganese as a luminescent center impurity, the
7 difference between the lattice constant, before lamination,
8 of the compound of the first layer and the compound
9 constituting the second layer is 5% or more, and the peak
10 value of the emission spectrum of the laminated luminescent
11 composite rests on 590 nm or more, whereby the thin-film EL
12 element can provide red light having high color purity.

U. S. Patent No. 5,635,110 teaches a multi-stage process for preparing a phosphor product which includes the stages of selecting precursors of a dopant and a host lattice as the phosphor starting materials, grinding the starting materials in an initial grinding stage for an initial grinding time period to produce an initial ground

1 material having a smaller particle size distribution than
2 the starting materials, firing the initial ground material
3 in an initial firing stage at an initial firing temperature
4 for an initial firing time period to produce an initial
5 fired material, grinding the initial fired material in an
6 intermediate grinding stage for an intermediate grinding
7 time period to produce an intermediate ground material
8 having a smaller particle size than the initial fired
9 material, wherein the intermediate grinding time period is
10 substantially less than the initial grinding time period,
11 firing the intermediate ground material in an intermediate
12 firing stage at an intermediate firing temperature for an
13 intermediate firing time to produce an intermediate fired
14 material, wherein the intermediate firing temperature is
15 substantially greater than the initial firing temperature,
16 grinding the intermediate fired material in a final grinding
17 stage for a final grinding time period to produce a final
18 ground material having a smaller particle size than the

1 intermediate fired material, and firing the final ground
2 material in a final firing stage at a final firing
3 temperature for a final firing time to produce a phosphor
4 product, wherein the final firing time is substantially less
5 than the intermediate firing time.

6 U. S. Patent No. 5,625,255 teaches an inorganic thin
7 film EL device which includes a substrate, a pair of
8 electrode layers and a pair of insulating layers formed on
9 the substrate in this order, and a light emission layer
10 sandwiched between the paired insulating layers and arranged
11 such that light emitted from the light emission layer is
12 taken-out from one side the light emission layer. The light
13 emission layer is made of a composition which consists
14 essentially of a fluoride of a metal of the group II of the
15 Periodic Table and a member selected from the group
16 consisting of rare earth elements and compounds thereof.
17 The metal fluoride is of the formula, $M_{1-x}F_{2+y}$ or
18 $M_{1+x}F_{2-y}$, wherein M represents a metal of the

1 group II of the Periodic Table, x is a value ranging from
2 0.001 to 0.9 and y is a value ranging from 0.001 to 1.8. The
3 device is useful as a flat light source.

4 U. S. Patent No. 5,621,069 teaches a technique for the
5 preparation of conjugated arylene and heteroarylene vinylene
6 polymers by thermal conversion of a polymer precursor
7 prepared by reacting an aromatic ring structure with an
8 aqueous solution of an alkyl xanthic acid potassium salt. In
9 this processing sequence the xanthate group acts as a
10 leaving group and permits the formation of a prepolymer
11 which is soluble in common organic solvents. Conversion of
12 the prepolymer is effected at a temperature ranging from 150
13 to 250 degrees C in the presence of forming gas. Studies
14 show that electro-luminescent devices prepared in accordance
15 with the described technique evidence internal quantum
16 efficiencies superior to those of the prior art due to the
17 presence of pinhole free films and therefore permit the
18 fabrication of larger area LED's than those prepared by

1 conventional techniques.

2 U. S. Patent No. 5,612,591 teaches an electro-
3 luminescent device which includes the sequential lamination
4 of a first electrode, first insulating layer, phosphor
5 layer, second insulating layer and second electrode while
6 using an optically transparent material at least on the side
7 on which light leaves the device; wherein, in. addition to
8 the phosphor layer being composed of calcium thiogallate
9 (CaGa.sub.2 S.sub.4) doped with a luminescent center
10 element, the host of the phosphor layer is strongly oriented
11 to the (400) surface.

12 U. S. Patent No. 5,608,287 teaches an electro-
13 luminescent device. The device has a bottom electrode layer
14 disposed on a substrate for injecting electrons into an
15 organic layer, and a top electrode, such as ITO, disposed on
16 the organic layer for injecting holes into the organic
17 layer. The bottom electrode is formed of either metal
18 silicides, such as, rare earth silicides, or metal borides,

1 such as lanthanum boride and chromium boride having a work
2 function of 4.0 eV or less. The electrodes formed from
3 either metal silicates, or metal borides provide protection
4 from atmospheric corrosion.

5 U. S. Patent No. 5,640,398 teaches an electro-
6 luminescence light-emitting device for generating an optical
7 wavelength which includes a substrate; an ITO layer coated
8 on the substrate, at least two light-emitting layers
9 sequentially formed on the ITO layer and having a different
10 band gap, and a metal electrode formed on an upper light-
11 emitting layer of the at least two light-emitting layers.
12 The ITO layer is used as an anode and the metal electrode is
13 used as a cathode.

14 U. S. Patent No. 5,598,059 teaches an AC thin film
15 electro-luminescent (TFEL) device which includes a multi-
16 layer phosphor for emitting white light having improved
17 emission intensity in the blue region of the spectrum. The
18 multi-layer stack consists of an inverted structure thin

1 film stack having a red light emitting manganese doped zinc
2 sulfide (ZnS:Mn) layer disposed on a first insulating layer;
3 a blue-green light emitting cerium doped strontium sulfide
4 (SrS:Ce) layer disposed on the red light emitting layer; and
5 a blue light emitting cerium activated thiogallate phosphor
6 (Sr.sub.x Ca.sub.1-x Ga.sub.2 S.sub.4 :Ce) layer disposed on
7 the blue-green light emitting layer. The manganese doped
8 zinc sulfide layer acts as a nucleating layer that lowers
9 the threshold voltage, and the cerium activated thiogallate
10 phosphor layer provides a moisture barrier for the
11 hygroscopic cerium doped strontium sulfide layer. The white
12 light from the multi-layer phosphor can be appropriately
13 filtered to produce any desired color.

14 U. S. Patent No. 5,593,782 teaches encapsulated
15 electro-luminescent phosphor particles. The particles are
16 encapsulated in a very thin oxide layer to protect them from
17 aging due to moisture intrusion. The particles are
18 encapsulated via a vapor phase hydrolysis reaction of oxide

1 precursor materials at a temperature of between about 25 to
2 about 170 degrees C., preferably between about 100 and about
3 150 degrees C. The resultant encapsulated particles exhibit
4 a surprising combination of high initial luminescent
5 brightness and high resistance to humidity-accelerated
6 brightness decay.

7 U. S. Patent No. 5,578,379 teaches siloxene and
8 siloxene derivatives. These derivatives are compatible with
9 silicon and which may be generated as epitaxial layer on a
10 silicon mono-crystal. This permits the production of novel
11 and advantageous electro-luminescent devices, such as
12 displays, image converters, optical-electric integrated
13 circuits. Siloxene and siloxene derivatives may also be
14 advantageously employed in lasers as laser-active material
15 and in fluorescent lamps or tubes as luminescent material.

16 U. S. Patent No. 5,574,332 teaches a low-pressure
17 mercury discharge lamp which includes a luminescent screen.
18 The luminescent screen includes a zeolite containing

1 trivalent Ce. The luminescent screen exhibits a large
2 quantum efficiency for converting W radiation of 254 nm into
3 radiation having an emission maximum at approximately 346
4 nm.

5 U. S. Patent No. 5,561,304 teaches an electro-
6 luminescent silicon device which includes a silicon
7 structure. The structure has a bulk silicon layer- and a
8 porous-silicon layer. The-porous layer has merged pores.
9 The pores define silicon quantum wires. The quantum wires
10 have a surface passivation layer. The porous layer exhibits
11 photoluminescence under ultra-violet irradiation. The
12 porous layer is pervaded by a conductive material such as an
13 electrolyte or a metal. The conductive material ensures
14 that an electrically continuous current path extends through
15 the porous layer; it does not degrade the quantum wire
16 surface passivation sufficiently to render the quantum wires
17 non-luminescent, and it injects minority carriers into the
18 quantum wires. An electrode contacts the conductive material

1 and the bulk silicon layer has an Ohmic contact. When
2 biased the electrode is the anode and the silicon structure
3 is the cathode. Electro-luminescence is then observed in
4 the visible region of the spectrum.

5 U. S. Patent No. 5,554,911 teaches a multi-color light-
6 emitting element which has at least two optical micro-cavity
7 structures having respectively different optical lengths
8 determining their emission wavelengths. Each micro-cavity
9 structure contains a film of or organic material as a light-
10 emitting region, which may be a single film of uniform
11 thickness in the element.

12 U. S. Patent No. 5,554,449 teaches a high luminance
13 thin-film electro-luminescent device which includes a
14 phosphor layer having SrS as the host material and a
15 luminous center. The phosphor layer is sandwiched between
16 two insulating layers and two thin-film electrodes are
17 provided on each side of the insulating layers. At least one
18 of the electrodes is transparent, and the excitation

1 spectrum of the phosphor layer exhibits a peak having a
2 maximum value at a wavelength of about from 350 nm to 370
3 nm. Such a high luminance thin-film electroluminescent
4 device can be prepared by annealing its phosphor layer
5 having SrS as the host material at a temperature of at least
6 650 degrees C for at least one hour in an atmosphere of a
7 sulfur-containing gas.

8 U. S. Patent No. 5,543,237 teaches an inorganic thin
9 film EL device which includes, on an insulating substrate, a
10 back electrode, an insulating layer, a light emission layer,
11 an insulating layer and a transparent electrode formed on
12 the substrate in this order. The emission layer includes a
13 fluoride of an alkaline earth metal and at least one member
14 selected from the group consisting of rare earth element
15 metals and compounds thereof at a mixing ratio by weight of
16 10:90 to 95:5. The rare earth element is, for example,
17 cerium, praseodymium, neodymium, samarium, europium,
18 gadolinium, terbium, dysprosium, holmium, erbium, thulium,

1 ytterbium and mixture thereof. The compounds may be those
2 compounds of the rare earth elements and fluorine, chlorine,
3 bromine, iodine and oxygen.

4 U. S. Patent No. 5,541,012 teaches a new infrared-to-
5 visible up-conversion material which can be applied to an
6 infrared light identification element having a useful
7 conversion efficiency and sensitivity for infrared light in
8 the wavelength of 1.5 micron band, 0.98 micron band and 0.8
9 micron band without the necessity of previous excitation of
10 the material. This infrared-to-visible up-conversion
11 material consists of an inorganic material comprising at
12 least two elements of erbium (Er) and a halogen or compounds
13 thereof.

14 U. S. Patent No. 5,540,999 teaches an electro-
15 luminescent element. The element includes an organic
16 compound layer formed of a thiophene polymer as a light
17 emitting layer or a hole-injection transport layer. The
18 element emits light at high luminance and is reliable.

1 U. S. Patent No. 5,536,588 teaches an amorphous organic
2 thin-film element containing dye molecules with
3 $\Delta S_{tr,m} (J/(K.kmol))/Mw$ of 60 or less, assuming
4 that the molecular weight is Mw and the sum total of an
5 entropy change of melting and entropy changes of transition
6 from a glass transition point to a melting point is
7 $\Delta S_{tr,m} (J/(K.kmol))$, and having a high heat
8 resistance and a high stability over long periods of time.

9 U. S. Patent No. 5,529,853 teaches an organic EL
10 element which includes a hole-injecting electrode and an
11 electron-injecting electrode, and at least a film made of a
12 luminous material there-between. The luminous material is
13 one of a metal complex polymer, an inner complex salt having
14 two or more ligands, and 10-hydroxybenzo [h] quinoline-metal
15 complex.

16 U. S. Patent No. 5,521,465 teaches an AC thin film
17 electro-luminescent display panel includes a metal assist
18 structure formed on and in electrical contact over each

1 transparent electrode and light absorbing darkened rear
2 electrodes. The electrodes combine to provide a sunlight
3 viewable display panel.

4 U. S. Patent No. 5,517,080 teaches an AC thin film
5 electro-luminescent display panel includes a metal assist
6 structure formed on and in electrical contact over each
7 transparent electrode, and a graded layer of light absorbing
8 dark material which combine to provide a sunlight viewable
9 display panel.

10 U. S. Patent No. 5,516,577 teaches an organic electro-
11 luminescence device which includes laminating layers in the
12 order of anode/light emitting layer/ adhesive layer/
13 cathode, or anode/hole-injecting layer/light emitting
14 layer/adhesive layer/cathode, the energy gap of the light
15 emitting layer being larger than that of 8-hydroxyquinoline
16 or metal complex thereof and contained in the adhesive
17 layer, the light emitting layer comprising a compound which
18 emits a blue, greenish blue or bluish green light in CIE

1 chromaticity coordinates, and the adhesive layer including a
2 metal complex of 8-hydroxyquinoline or a derivative thereof
3 and at least one organic compound in an arbitrary region in
4 the--direction of the thickness of the layer, the thickness
5 of which is smaller than that of the above-mentioned light
6 emitting layer. According to the above organic electro-
7 luminescence device, improvements in uniformity in light
8 emission and emission efficiency are realized.

9 U. S. Patent No. 5,508,585 teaches an EL lamp includes
10 a transparent electrode, an electro-luminescent dielectric
11 layer overlying the transparent electrode, a patterned
12 insulating layer overlies selected portions of the
13 dielectric layer for reducing the electric field across the
14 selected portions of the electro-luminescent dielectric
15 layer, and a rear electrode overlying the insulating layer
16 and the electro-luminescent dielectric layer. The insulating
17 layer is preferably a low dielectric constant material and
18 can overlie the electro-luminescent dielectric layer or can

1 be located between a separate dielectric layer and a
2 phosphor layer. A gray scale is produced by depositing or
3 printing more than one thickness of insulating layer.

4 U. S. Patent No. 5,500,568 teaches an organic EL device
5 having, as a cathode, a vapor deposited film containing at
6 least one metal A selected from Pb, Sn and Bi and a metal B
7 having a work function of 4.2 eV or less has high chemical
8 stability of the cathode with time and high power conversion
9 efficiency, and is useful as a display device and a light-
10 emitting device.

11 U. S. Patent No. 5,491,377 teaches a flexible, thick
12 film, electro-luminescent lamp in which a single non-
13 hygroscopic binder is used for all layers (with the optional
14 exception of the rear electrode) thereby reducing
15 delamination as a result of temperature changes and the
16 susceptibility to moisture. The binder includes a fluoro-
17 polymer resin, namely poly-vinylidene fluoride, which has
18 ultraviolet radiation absorbing characteristics. The use of

1 a common binder for both phosphor and adjacent dielectric
2 layers reduces lamp failure due to localized heating, thus
3 increasing light output for a given voltage and excitation
4 frequency, and increasing the ability of the lamp to
5 withstand over-voltage conditions without failure. The
6 lamps may be made by screen-printing, by spraying, by roller
7 coating or vacuum deposition, although screen printing is
8 preferred. By the multi-layer process, unique control of
9 the illumination is achieved.

10 U. S. Patent No. 5,487,953 teaches an organic electro-
11 luminescent device which includes an organic emitting layer
12 and a hole-transport layer laminated with each other and
13 arranged between a cathode and an anode, in characterized in
14 that the hole transport layer made of the triphenylbenzene
15 derivative. This hole-transport layer has the high heart-
16 resistant property and high conductivity to improve the
17 durability and thus this device emits light at a high
18 luminance and a high efficiency upon application of a low

1 voltage.

2 U. S. Patent No. 5,484,922 teaches an organic electro-
3 luminescent device which employs, an aluminum chelate of the
4 formula: wherein n is 1 and x is 1 or 2, or n is 2 and x is
5 1; and, Q is a substituted 8-quinolinolato group in which
6 the 2-position substituent is selected from the group
7 consisting of hydrocarbon groups containing from 1 to 10
8 carbon atoms, amino, aryloxy and alkoxy groups; L is a
9 ligand, each L ligand being individually selected from (a)
10 the group consisting of --R, --Ar, --OR, --ORAr, --OAr, --
11 OC(0)R, --OC(0)Ar, --OP(0)R.sub.2, --OP(0)Ar.sub.2, --
12 OS(O.sub.2)R, --OS(O.sub.2)Ar, --SAr, --SeAr, --TeAr, --
13 OSiR.sub.3, --OSiAr.sub.3, --OB(OR).sub.2, --OB(OAr).sub.2,
14 and --X, when x is 1, or from (b) --OC(0)Ar'C(0)O-- or --
15 OAr'O--, when x is 2, where R is a hydrocarbon group
16 containing from 1 to 6 carbon atoms, Ar and Ar' are,
17 respectively, monovalent and divalent aromatic groups
18 containing up to 36 carbon atoms each, and X is a halogen;

1 with the proviso that when L is a phenolic group n is 2 and
2 x is 1.

3 U. S. Patent No. 5,456,988 teaches an electro-
4 luminescent device having a hole injection electrode, an
5 electron injection electrode, and at least an organic
6 emitting layer there-between. The organic emitting layer
7 includes an 8-quinolinol derivative-metal complex whose
8 ligand is selected from the group consisting of chemical
9 formulas 102 through 106: chemical formula 102 ##STR1##
10 chemical formula 103 ##STR2## chemical formula 104 ##STR3##
11 chemical formula 105 ##STR4## chemical formula 106 ##STR5##.

12 U. S. Patent No. 5,453,661 teaches a flat panel display
13 which includes a ferro-electric thin film between the first
14 and second spaced apart electrodes. The ferro-electric thin
15 film emits electrons upon application of a predetermined
16 voltage between the first and second spaced apart
17 electrodes. The electrons are emitted in an electron
18 emission path and impinge upon a luminescent layer such as a

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phosphor layer, which produces luminescence upon impingement upon the emitter electrodes. The ferro-electric thin film is preferably about 2 microns or less in thickness and is preferably a polycrystalline ferro-electric thin film. More preferably, the thin ferro-electric film is a highly oriented, polycrystalline thin ferro-electric film. Most preferably, highly oriented ferro-electric thin film has a preferred (001) crystal orientation and is about 2 microns or less in thickness. A flat panel display may be formed of arrays of such display elements. Top and bottom electrodes or side electrodes may be used. The display may be formed using conventional microelectronic fabrication steps.

U. S. Patent No. 5,449,564 teaches an EL element which has at least one layer made from an organic material between an electron injection electrode and a hole injection electrode. The organic material consists of an oxadiazole series compound. The compound has a plurality of oxadiazole rings. Each oxadiazole ring is substituted by a condensed

1 polycyclic aromatic group.

2 U. S. Patent No. 5,444,268 teaches a thin film EL
3 device.

4 U. S. Patent No. 5,443,922 teaches an organic thin film
5 electro-luminescence element.

6 U. S. Patent No. 5,443,921 teaches a thin film electro-
7 luminescence device.

8 U. S. Patent No. 5,442,254 teaches a fluorescent device
9 with a quantum contained particle screen.

10 U. S. Patent No. 5,432,014 teaches an organic electro-
11 luminescent element.

12 U. S. Patent No. 5,429,884 teaches an organic electro-
13 luminescent element.

14 U. S. Patent No. 5,405,710 teaches an article including
15 micro-cavity light sources.

16 U. S. Patent No. 5,404,075 teaches a TFEL element with
17 a tantalum oxide and a tungsten oxide-insulating layer.

18 U. S. Patent No. 5,400,047 teaches a high brightness

1 thin film electro-luminescent display with low OHM
2 electrodes.

3 U. S. Patent No. 5,382,477 teaches an organic electro-
4 luminescent element.

5 U. S. Patent No. 5,374,489 teaches an organic electro-
6 luminescent device.

7 U. S. Patent No. 5,336,546 teaches an organic electro-
8 luminescence device.

9 U. S. Patent No. 5,328,808 teaches an edge emission
10 type electro-luminescent device arrays.

11 U. S. Patent No. 5,320,913 teaches conductive film and
12 low reflection conductive film.

13 U. S. Patent No. 5,319,282 teaches a planar fluorescent
14 and electro-luminescent lamp having one or more chambers.

15 U. S. Patent No. 5,314,759 teaches a phosphor layer of
16 an electro-luminescent component.

17 U. S. Patent No. 5,311,035 teaches a thin film electro-
18 luminescence element.

1 U. S. Patent No. 5,309,071 teaches zinc sulfide
2 electro-luminescent phosphor particles and electro-
3 luminescent lamp made therefrom.

4 U. S. Patent No. 5,309,070 teaches a TFEL device having
5 blue light emitting thiogallate phosphor.

6 U. S. Patent No. 5,306,572 teaches EL element which has
7 an organic thin film.

8 U. S. Patent No. 5,300,858 teaches a transparent
9 electro-conductive film, an AC powder type EL panel and a
10 liquid crystal display using the same.

11 U. S. Patent No. 2,445,692 teaches an ultraviolet lamp.

12 U. S. Patent No. 2,295,626 teaches an ultraviolet lamp.

13 U. S. Patent No. 3,845,343 teaches a bulb for an
14 ultraviolet lamp.

15 The inventor hereby incorporates the above patents by
16 reference.

17 SUMMARY OF THE INVENTION

18 The present invention is directed to a biochip which

1 has a sensor.

2 In a first aspect of the invention the sensor contains
3 a light source and an optical detector.

4 In a second aspect of the invention the light source
5 is an electro-luminescent material.

6 Other aspects and many of the attendant advantages will
7 be more readily appreciated as the same becomes better
8 understood by reference to the following detailed
9 description and considered in connection with the
10 accompanying drawing in which like reference symbols
11 designate like parts throughout the figures.

12 The features of the present invention which are
13 believed to be novel are set forth with particularity in the
14 appended claims.

15 DESCRIPTION OF THE DRAWINGS

16 Fig. 1 is a schematic drawing of a 1.0" x 1.0" optical
17 array of 90 dye doped porous silica microspheres.
18 Represented are three fluorescent dyes: fluorescein,

1 coumarin, and rhodamine-B. Viewed under 365 nm W
2 excitation.

3 Fig. 2 is a schematic representation of a multiple dye
4 doped porous silica microsphere for sensing applications.
5 Microsphere diameters range from 500 nm to 2.0 mm, with pore
6 diameters ranging from 1.7 nm to 100 nm.

7 Fig. 3 is a schematic drawing of three fluorescent dye
8 doped porous silica microsphere sensors with 365nm
9 excitation (up through large diameter plastic waveguide).

10 Fig. 4 is an example of a multi-microsphere sensor
11 employing hexavalent urania doped porous silica.

12 Fig. 5 is a schematic drawing of the ratio
13 (525nm/475nm) of fluorescent emission of fluorescein doped
14 porous silica microspheres excited at 365 nm. Equilibrium
15 time approximately 2 minutes.

16 Fig. 6 is schematic drawing of an example of a single
17 sensor element from a MEMs based sensor array using porous,
18 dye/protein doped silica microspheres.

Fig 7 through Fig. 22 are schematic drawings of alternative designs including one design which involves "V" shaped troughs with the EL material on one face and the silicon based photodetector on the other with dye-doped and optically active protein doped porous gel microspheres filling the trough.

DESCRIPTION OF THE PREFERRED EMBODIMENT

U. S. Patent 5,496,997 (March 5, 1996) teaches a sensor which incorporates an optical fiber and a solid porous inorganic microsphere and an optical fiber which having a proximal end and a distal end. The distal end of the optical fiber is coupled to the porous microsphere by an adhesive material. The porous microsphere is doped with a dopant. The dopant may be either an organic dye or an inorganic ion. A sensing apparatus includes the sensor, a spectrophotometer and a source of light. The spectrophotometer is coupled to the proximal end of the optical fiber. The source of light causes either the organic dye or the inorganic ion to

1 fluoresce.

2 Tremendous progress has been made in recent years in
3 understanding some of the fundamental aspects of chemical
4 and biological sensing. Most research and commercialization
5 efforts have been focused upon fabricating individual
6 sensors for specific and usually narrow applications and
7 application environments. An excellent overview of the
8 subject emphasizing both the challenges and commercial
9 opportunities is given by Weetal [1]. Inasmuch as most
10 commercially available chemical and biological sensors were
11 developed independently of one another, trying to integrate
12 them into one device would be extremely difficult and
13 costly. The challenge of integration rests primarily on
14 developing a multifunctional "platform" sensing technology
15 that can allow the high volume, low cost fabrication of
16 large numbers of individual sensors on a single array. Just
17 as an image on a view screen is composed of a large number
18 of light generating pixels, a sensor array would also be

1 composed of a large number of "sensels", individual sensor
2 elements to generate an "image" or map of an unknown
3 substance, be it liquid or vapor, being examined. Emphasis
4 needs to be given to the types of platform approaches that
5 have the greatest likelihood of supporting broad based
6 sensing capabilities. Traditional gas sensor technologies,
7 as an example, offer little hope of this type of broad
8 sensing capability [2].

9 This is not a comprehensive review, but an overview of
10 some of the most exciting recent developments made by
11 researchers in the field that point to an approach that
12 could provide a broad based sensing platform. It also sets
13 the stage for our proposed sensor technology, MEMOSA, which
14 stands for MEMs based Optical Sensor Array. Through the
15 merging of technologies and resources from both MATECH and
16 several university and industry collaborators, highly
17 sophisticated, commercially viable sensor systems could be
18 practical within only a few years.

Integrated sensor arrays permit a single platform for a wide range of simultaneous sensing operations to be conducted. Both optically and electronically based array systems are possible and have been recently demonstrated. In an early example, light to be measured from an unknown source can be passed through a diffraction grating and on to an array of sensors [3]. In this manner, solid-state spectrophotometers using optical fibers to conduct the light from an unknown source can be constructed. By incorporating the chemically and/or biologically active components on to a array of photodiodes and/or electrodes, more sophisticated sensor arrays can be fabricated. Three examples of integrated sensor arrays are highlighted in this section.

Rapidly and accurately detecting fragments of DNA is critically important for the clinical diagnosis of a wide range of genetically predetermined disease states. By detecting the genetic markers of diseases before they become outwardly manifest, allows early intervention and treatment.

1 DNA markers can also signal the initial metastasis of a
2 wide number of cancers. Current hybridization methods
3 typically require high sample DNA concentration for accurate
4 analyses [4]. In vitro amplification technologies, such as
5 PCR require lengthy assay times in order to overcome this
6 problem. Several researchers have pioneered novel
7 approaches to achieve rapid and highly sensitive DNA
8 detection. Ferguson and co-workers have demonstrated a
9 fiber-optic DNA biosensor array with a bundle of seven (7)
10 fibers in a small probe [4]. The only significant drawback
11 is that labelled sample targets were required [4].
12 Affymetrix (Santa Clara, CA) has recently demonstrated a DNA
13 chip with 12,224 different oligonucleotide probes [5,6]. A
14 key drawback to their technology is that "the chips only
15 read what they are designed to read - you have to know a
16 reference sequence beforehand to design probes to detect
17 variations in that sequence"[5]. Research is also focused on
18 designing better optical probes [7-9]. Recent research at

1 the Public Health Research Institute has shown that using
2 "hairpin shaped oligonucleotide probes" greatly enhances
3 specificity [6]. As originally predicted by Leroy Hood and
4 co-workers in 1988, the tremendous progress in deciphering
5 the human genome, coupled with advances in diagnostic
6 technology could result in a revolutionary advance in
7 disease detection and diagnosis [10].

8 Another sensor array area which has shown great
9 commercial promise just recently is the effort to develop an
10 "artificial nose". The science of how we smell is extremely
11 complex [11]. Recent progress has been achieved in mapping
12 how the olfactory system operates [12]. In a recent movie,
13 "Richie Rich", a comedy shows research scientists developing
14 a hand held device called the SMELL MASTER 2000, which can
15 discriminate between a fine merlot and a cheap jug wine!
16 Unfortunately, the technological challenges make that kind
17 of sensitivity still a fantasy. A recent effort to model a
18 sensor system after the vertebrate olfactory system has been

demonstrated by Dickenson, et al.[13]. They use a multitude of dye doped polymers at the end of optical fibers to form a fluorescent response pattern to specific analytes. By employing a distributed sensing approach, they must "train" a neural network for specific vapor recognition [13]. Once they have a pattern or signature for each compound, then the "sniffer" can recognize it if it "smells" it again. One of the drawbacks of this approach is trying to discriminate between complex mixtures of vapors. Another similar approach, but using the electrical properties of an array of 16 carbon black doped porous polymers is being pursued by Cyranno Sciences (Pasadena, CA). Their patented technology, licensed from CALTECH, permits a 3-dimentional odor map to be created based upon the response of the sensor array for a wide variety of "smells"[14]. Instead of trying to analyze the constituent components of an odor, they focus upon its overall or composite smell. In this manner, they may actually be able to distinguish between a cabernet and a

1 merlot! But I'd rather do that job myself.

2 Optical sensor arrays can be fabricated by coupling an
3 array of dye/protein doped microspheres to individual
4 optical optical fibers which can be multiplexed into a
5 spectrophotometer. Linear arrays of optical fibers are now
6 commercially employed in DNA sequence detectors and
7 fluorescence based microtiter plate readers used for ELISA
8 tests in clinical diagnostics. An example of a linear array
9 of optical fibers appears in the Perkin Elmer Applied
10 Biosystems 7700 DNA sequence Analyzer (Foster City, CA).
11 The approach can be augmented by the attachment of
12 fluorescence based sensors in the form of microspheres,
13 doped with chemically or biologically active reporter
14 molecules (see sections 5.2 and 5.3).

15 Referring to Fig. 1 a two dimensional array of 90
16 porous, dye-doped silica microspheres in which three types
17 of dye-doped spheres are alternated in a repeated pattern.

18 For any sensor array system, pattern recognition

protocols are critical. In the two previous examples, DNA sensors and the artificial nose, data from the sensor arrays must be analyzed to "interpret" the pattern of signal from the individual sensor cells that make up the total array. This "intelligence" is not unlike that required for pattern recognition systems currently used for both military, law enforcement and commercial systems designed to recognize shape or morphology, such as the profile of a tank, the unique pattern of a fingerprint, or the shape and size of potato. Behind the architecture of data collection must reside a logic-software to maximize the efficiency of pattern recognition. Usually, these logic loops are hierarchical in nature [15].

A simple example, taking from everyday life, is how one recognizes his mom's sport utility vehicle (SUV). Both his parents and he live in the same town, so he is accustomed to seeing them periodically while driving. It takes only a split second to complete the five step process (were it

1 otherwise he might run into someone). First, he notices the
 2 shape (a typical SUV). Then the color (black). Third, he
 3 looks for a spare tire attached to the back (there shouldn't
 4 be one). Next come the door handles (the back door handles
 5 should be on the side of the rear window). Finally, he
 6 looks to recognize the occupants (his mom and his dad?). By
 7 truncating my analysis at one of the earlier steps, he can
 8 shorten the time required to rule-out the suspect vehicle as
 9 belonging to his parents. If he closely studies the
 10 occupant of every car on the road, he surely be a public
 11 menace! Having well designed logic loops for screening
 12 while using a sensor array can accelerate the speed of
 13 operation of sensor systems. Integrating the sensing system
 14 with data collection and interpretation (i.e. software) is
 15 necessary for an efficient sensor system.

16 Fiber-optic sensing has emerged in recent years as a
 17 powerful tool for the development of "smart systems".
 18 Applications include medical diagnostics, environmental

1 testing, and industrial monitoring. Optical fibers can be
2 deployed across large distances, often to remote locations
3 which are difficult or impossible to access by other means.
4 Fibers can be used for medical biopsies of the human body,
5 sent down wells, mine shafts, or to the bottom of lakes,
6 rivers and streams. To date, however, fiber-optic sensing
7 has been limited to only few narrowly defined applications.
8 In order to fully exploit the potential of optical fibers
9 for sensing applications, a new, more versatile platform
10 technology is needed.

11 Jane and Pinchuk teach a method of fabricated fiber-
12 optic chemical sensors using charged hydrogel matrices for
13 the immobilization of colorimetric indicators for the
14 measurement of pH and other applications [16]. Using the
15 phenomenon of thermo-luminescence, Kera, et al teach the
16 method of high temperature flame detection and monitoring
17 employing lanthanide doped optical fibers [17]. Grey et al
18 have shown a system based upon dual fiber optic cells for

1 serum analysis [18]. Wixom teaches a method of shock
2 detection based upon electroluminescent optical fibers [19].
3 Kane has demonstrated measuring both blood pH and oxygen
4 levels using fiber optic probes [20]. Fiber optic carbon
5 dioxide sensors have been developed for monitoring
6 fermentation processes [21]. Immunosensors based upon
7 enhanced chemoluminescence and fiber optics have also been
8 demonstrated [22].

9 Employing the sol-gel route, porous glass microspheres,
10 doped with a wide range of optically-active organic and
11 inorganic molecules have been demonstrated [23,24]. It has
12 also been demonstrated that a glass microsphere can be
13 mounted to the end of an optical fiber as a lens [25]. By
14 attaching a dye-doped porous microsphere to the end of an
15 optical fiber, a versatile new sensor system has been
16 developed [26,27]. More about these new sensors is
17 described in the following section.

18 An alternative approach, pursued by most researchers in

1 the field, is using gel encapsulation to immobilize dyes,
2 proteins, enzymes, and antibodies as part of a thin cladding
3 on a length of the optical fiber [28-30]. This relies upon
4 the evanescent field effect, thereby requiring a certain
5 length of fiber for sensing to be sensitive. Advantages of
6 this method include fast response time. A major
7 disadvantage is that a significant length of fiber is
8 usually needed (at least a few cms) for sensitivity. Others
9 have examined using a small "monolith" of gel encapsulated
10 material at the end of an optical fiber [31]. The potential
11 for using high surface area gel encapsulated antibodies has
12 not been realized inasmuch as the typical pore sizes of
13 silica gels is smaller than the size of the pathogens being
14 detected. Nonetheless, Ligler and colleagues have
15 demonstrated, by conjugating antibodies to the outer surface
16 of an optical fiber, that this type of biosensing has great
17 potential utility [32]. The encapsulating of antibodies in
18 a host of high pore volume and large surface area might

1 result in much greater sensitivity. Materials potentially
2 suitable for such an application are described in the
3 following section.

4 Unlike traditional glass and ceramic processing
5 methods, in which powdered oxides are heated to high
6 temperatures, the sol-gel process permits the fabrication of
7 inorganic gels at temperatures near ambient from liquid
8 solutions [33]. Avnir and co-workers were the first to
9 demonstrate the possibility of incorporating optically-
10 active organic dye molecules into porous gels [34]. More
11 recently, MacCraith and co-workers have successfully
12 demonstrated the possibility of fiber-optic sensing through
13 the application of dye-doped porous silica films to the end
14 of optical waveguides [35,36]. Their sensors take advantage
15 of evanescent wave interactions, such as evanescent wave
16 absorption and evanescent wave excitation of fluorescence
17 [35].

18 Referring to Fig. 2 dye-doped porous silica

1 microspheres have been prepared from liquid solutions [37].
2 A wide range of optically-active dopants have been
3 incorporated into silica microspheres, including both
4 organic and inorganic species [37]. Luminescent
5 microspheres have previously been demonstrated for flat-
6 panel display applications [38-40]. The incorporation of
7 dye-doped porous silica microspheres into a fiber-based
8 sensing system has been demonstrated by attaching a porous,
9 dye or protein doped microsphere to the distal end of an
10 optical fiber [26,27]. Ultraviolet or blue light can be
11 utilized to excite fluorescence of the optically-active dye
12 molecule.

13 Referring to Fig. 3 in conjunction with Fig. 4 three
14 microspheres, doped with fluorescein, coumarin, and
15 rhodamine-B, are shown each attached to an optical fiber in
16 under UV excitation. A wide range of prototype sensors
17 based upon multiple doped microspheres have been developed.

18 MATECH announced the availability of a series of new,

1 highly porous silica supports for liquid chromatography,
2 catalysis, biosensing, and protein separation applications.
3 MATECH's range of large pore materials represent the first
4 commercial availability of porous silica that possesses both
5 large pore diameters and large pore volumes, attributes
6 critical to large protein and monoclonal antibody
7 separations, for example. While preserving high pore
8 volumes, MATECH's new line of materials have pore sizes
9 ranging from 1.7 to 100 nanometers (17 - 1000 angstroms).A
10 complete list of MATECH's new line of materials is listed
11 below.

12 MATERIAL

13 TYPE PORE SIZE

14 (Angstroms) SURFACE AREA

15 (m²/gm) PORE VOLUME

16 (cc/gm)

17

18 A 17 400 0.3

1 B 100 500 0.7
2 C 160 900 2.2-3.0
3 D 250 1100 2.2-3.0
4 E 500 450 2.0-3.0
5 F 1000 400 1.5-2.0

6 Lucan and co-workers have demonstrated the use of
7 fluorescein dye in sol-gel thin films for possible pH
8 measurement applications [41]. In their work, changes in
9 the absorption spectra of the fluorescein dye molecule after
10 immersion in aqueous solutions of various pH values were
11 measured. Repeat cycles were demonstrated. More recently,
12 evanescent excitation of fluorescein emission in a doped
13 thin film clad region of a 7 meter optical fiber pH sensor
14 has been shown [42].

15 In the inventor's previously published work,
16 fluorescein-doped porous silica microspheres were immersed
17 in aqueous solutions of various pH values[26]. The
18 fluorescence emission, after a few minutes of immersion, was

1 measured. A significant variation in the fluorescent
2 emission, particularly for pH values between 1 and 7, were
3 observed.

4 Referring to Fig. 5 the change in the ratio of
5 fluorescence emission at 475 and 525 nm is plotted vs. pH
6 value.

7 The use of 8-hydroxy-1,3,6-pyrenetrisulfonic acid
8 trisodium salt, "pyranine", as a sensitive molecular probe
9 for measuring alcohol content of gels has been demonstrated
10 [43,44]. More recently, the staining of microorganisms with
11 pyranine dye prior to gel encapsulation as a biological
12 probe has been performed on *S. cerevisiae* to monitor ethanol
13 evolution during fermentation [45,46]. Pyranine readily
14 exists in a protonated and deprotonated state. The
15 protonated pyranine fluoresces at 430 nm and the
16 deprotonated pyranine fluoresces at 515 nm. Initially, the
17 pyranine in dried silica gel is fully protonated. After
18 immersion in 0.1 M NH_4OH solution, pyranine becomes fully

1 deprotonated. . Switching protonation states has been
2 demonstrated to be fully reversible. By immersing pyranine-
3 doped silica microspheres in solutions of ethanol and
4 buffered water of varying alcohol contents, the ratio of
5 protonated to deprotonated fluorescence could be obtained
6 and plotted [26].

7 It is well known that the fluorescence behavior of
8 organic dye molecules is sensitive to temperature effects in
9 solution, particularly for dye laser applications[47].
10 Organic dyes, when incorporated into solid-state hosts,
11 should be expected to exhibit similar effects. The
12 fluorescence emission of fluorescein-doped silica
13 microspheres, measured at 0 C and 75 C has been previously
14 published [48]. Using organic dyes, a sensitive fiber-optic
15 thermometer should be possible for temperatures near
16 ambient. In recent unpublished work, the temperature
17 dependence of the fluorescent emission of hexavalent uranium
18 oxide doped silica gel beads or melt glass beads could

1 provide sensitive, optical temperature measurement
2 capabilities up to approximately 800oC.

3 The ability to detect even trace quantities of heavy
4 metals is of increasing importance for environmental
5 testing. It has long been known that heavy metals, such as
6 lead, form highly stable organometallic compounds [49].
7 Mackenzie and co-workers have recently shown that organic
8 molecules incorporated into gels and ORMOSILs can bond with
9 heavy metals, such as lead and hexavalent chromium,
10 contained in liquid solutions [50].

11 By doping silica gel with malachite green, Wong and
12 Mackenzie were able to measure hexavalent chromium in
13 aqueous solutions down to ~50 ppb[51]. The primary
14 mechanism of detection is based upon changes in the
15 absorption spectra of malachite green. By co-doping with a
16 fluorescent dye molecule, selected for an overlap between
17 the peak absorption of malchite green and the fluorescence
18 peak position of the luminescent dye molecule, it should be

1 possible to construct a fluorescence-based microsensor, as
2 well.

3 Malachite green is readily soluble in various silica
4 microsphere forming solutions [26]. In previously published
5 work, it has been shown that two prominent peaks in the
6 visible region of the absorption spectra are apparent, at
7 425 nm and 618 nm [26]. Moreover, it was shown that the
8 ratio of these peaks changes with exposure to hexavalent
9 chromium. By plotting the ratio of these peaks vs. Cr
10 concentration, a sensitive measurement system for Cr content
11 has been recently demonstrated.

12 Oka and Mackenzie have incorporated ethylene diamine
13 tetra-acetic acid (EDTA) into porous silica gels [50]. EDTA
14 is a well-known chelating agent for heavy metals [52].
15 Preliminary tests reveal it is possible to incorporate EDTA
16 into porous silica microspheres (about 1.0 gm) which, upon
17 exposure to 1.5 ml of lead solution (1000 ppm), result in a
18 measurable reduction (by ~50 percent) of lead (to about 500

1 ppm). The barely detectable fluorescence emission of EDTA
2 does change slightly in response to lead exposure.

3 Organophosphonates, such as PBTC and HEDP are widely
4 used for process control of water cooling towers, such as in
5 controlling corrosion and antiscaling. It has demonstrated
6 that fluorescent behavior of trivalent lanthanides, such as
7 cerium, terbium, and europium, in solution change upon
8 exposure to PBTC and HEDP. Unfortunately, a preliminary 9
9 month feasibility studied has shown that when bound into
10 porous silica gel support, any optical changes are not
11 easily measurable. Using other species, such as transition
12 metal ions (absorption) and actinides (hexavalent uranium),
13 however, rapid reversible sensors could be fabricated with
14 short response times (under two minutes). This is more than
15 adequate for heavily damped systems like water cooling
16 towers. More detailed results will be published in the near
17 future.

18 The first known disclosure of the incorporation of

1 organic proteins in silica gel was by Mackenzie and Pope
2 [53]. Braun and co-workers first demonstrated the ability to
3 incorporate enzymes in porous gels and show bio-reactivity
4 [54]. Ellerby et al. were able to demonstrate enzymatic
5 sensing using doped ORMOSILS [55]. Extensive progress in
6 understanding the fundamental science of biologically-active
7 proteins and enzymes in sol-gel silicates has occurred in
8 recent years [56-61]. The encapsulation of five analytical
9 coupling enzymes in silica microspheres by MATECH has been
10 described previously [26], but is repeated here for clarity.

11 These proteins and enzymes include R-phycoerythrin,
12 catalase, hexokinase, luciferase, and alcohol dehydrogenase.

13 R-phycoerythrin is one of several useful
14 phycobiliproteins derived from cyanobacteria and eukaryotic
15 algae[62]. This class of proteins is highly fluorescent and
16 has been conjugated with a wide range of antibodies and
17 compounds. The feasibility of doping silica gel and silica
18 microspheres with R-phycoerythrin has been demonstrated

1 [26,59]. The fluorescence spectra of R-phycoerythrin in
2 silica gel microspheres is virtually identical to that
3 obtained from R-phycoerythrin in aqueous solution [26]. The
4 incorporation of conjugated forms of this protein for
5 specific antibody and surface antigen sensing applications
6 holds great promise.

7 Catalase is well known to be an effective detector of
8 hydrogen peroxide. The photoluminescence spectra of
9 catalase-doped silica microspheres exposed to distilled
10 water and to 3% hydrogen peroxide solution has been
11 previously published. A pronounced shift in both intensity
12 and relative peaks heights of the two dominant peaks was
13 readily observed.

14 Continuous spectrophotometric rate determination is
15 utilized in the enzymatic assay of hexokinase for glucose
16 detection. The reaction path is as follows:
17 D-glucose + ATP -----(hexokinase)---? D-glucose 6-phosphate
18 + ADP D-glucose 6-phosphate + β -NADP -----(G-6-PDH)---? 6-PG

1 + β -NADPH where;

2 ATP = adenosine 5'-triphosphate,

3 ADP = adenosine 5'-diphosphate,

4 G-6-PDH = glucose-6-phosphate dehydrogenase,

5 β -NADP = β -nicotinamide adenine dinucleotide
6 phosphate, oxidized form,

7 β -NADPH = β -nicotinamide adenine dinucleotide
8 phosphate, reduced form.

9 Using these pathways, glucose detection can be measured
10 spectroscopically with high precision. The UV-vis-nIR
11 absorption spectra for hexokinase-doped silica gel has been
12 published previously [26]. Experiments to co-dope with ATP
13 and G-6-PDH and to explore alternate and reversible glucose
14 sensing pathways are the subject of in-house research.

15 ATP detection employing luciferin and luciferase
16 follows the reaction pathways, ATP + luciferin -----
17 (firefly luciferase)---? adenylyl-luciferin + PPi
18 adenylyl-luciferin + O₂ -----? Oxyluciferin + CO₂ +

1 light

2 The fluorescence spectra of firefly luciferase in
3 silica gel has been published previously [26]. The spectra
4 is identical to spectra obtained for luciferase in solution.
5 Moreover, recent unpublished results have shown that
6 bioluminescent spectra (assays) obtained when microspheres
7 co-doped with both luciferin and firefly luciferase are
8 exposed to ATP are identical to the photoluminescent
9 emission spectra. Conducting ATP assays at the end of an
10 optical fiber is completely feasible.

11 Bilirubin is the most significant constituent of bile
12 fluids secreted by the liver through the bile ducts into the
13 duodenum. It is a breakdown product of heme formed from the
14 degradation of erythrocyte hemoglobin in
15 reticuloendothelial cells, as well as other heme pigments,
16 such as cytochromes. Bilirubin is taken up in the liver and
17 conjugated to form bilirubin diglucuronide, which is
18 excreted in the bile. As an intensely colored (brown)

1 substance, its concentration in fluids can be readily
2 detected by spectrophotometric measurements (absorption).
3 Care, however, should be taken to eliminate any other
4 potential sources of absorption, such as bleeding ulcers and
5 food coloration. By "multipoint measurements" and patient
6 fasting, these two potential sources of interference might
7 be ruled out. While the fluorescent behavior of bilirubin
8 is less well understood, it may be possible to develop a
9 sensor for bilirubin based upon fluorescence, as well.
10 Using reflectance spectroscopy, bilirubin uptake within
11 porous silica beads may be possible, particularly if a
12 "porous mirror" can be deposited on the front end of the
13 bead (by physical vapor deposition PVD). An array of 90
14 hemi-spherically "mirrored" beads has already been
15 fabricated, demonstrating the possibility of the fabrication
16 process.

17 MATECH has already demonstrated the ability to
18 encapsulate fluorescent-labeled antibodies (fluorescein

1 tagged HIV antibody) in silica gel microporous beads for
2 surface antigen detection (HIV glycoprotein 120) [70]. We
3 propose to also evaluate the potential use of labelled
4 antibodies for the detection of legionella bacteria,
5 associated with recirculating water cooling systems and
6 airconditioning systems.

7 The inventor has evaluated the potential use of
8 labelled antibodies for the detection of H. pylori bacteria,
9 associated with ulcers and cancer. Labelled antibodies for
10 H. Pylori are already commercially available. The detection
11 strategy would be to determine spectroscopic changes (either
12 fluorescence or absorption) which occur when the conjugated
13 antibody comes in contact with the surface antigen (which is
14 continuously shed by the organism). Initial efforts could
15 be focused on simple "yes/no" detection. Future efforts
16 could focus on a more quantitative measurement of bacterial
17 concentration. While the bacteria is far too large to
18 penetrate the porous silica gel beads, the surface antigens

1 are very small. Researchers in France have shown that free
2 floating surface antigens, shed by their cells, can easily
3 diffuse into porous silica of a nominal 150 angstrom pore
4 diameter [63].

5 Living cells manifest a wide range of highly sensitive
6 metabolic processes and represent an opportunity to develop
7 highly sensitive biological sensors. Challenges to
8 developing whole cell based sensors include keeping them
9 alive and interfacing with the cell's metabolic functions.
10 Nonetheless, whole cell biosensing is emerging as an
11 exciting new area of research and development. The issue of
12 keeping the cells alive can be mitigated in in vivo sensing
13 applications. Palti has patented the use of living tissue
14 cells as sensors for blood and constituent levels, such as
15 glucose monitoring [64]. One drawback to in vivo
16 applications is the need to immunoisolate the foreign cells
17 to avoid immunorejection reactions. Researchers at Stanford
18 have already demonstrated how to make simple non-

1 immunoisolated sensors from living cells [65,66]. In their
2 work, they demonstrated ATP measurement and detection among
3 other things.

4 The issue of immunoisolation has been largely resolved
5 by our research into microbial and mammalian tissue cell
6 encapsulation [67-71]. While the bulk of our research,
7 which has now been spun-off into a separate company Solgene
8 Therapeutics, LLC, has been centered around biotech drug
9 delivery and cell therapy. For example, silica gel
10 encapsulated pancreatic islet allografts have been
11 successfully transplanted into severely diabetic mice,
12 resulting in a complete remission of symptoms (glucosuria
13 and high hematological glucose levels) for in excess of four
14 months [67,71]. No rejection of the encapsulated foreign
15 tissue was observed. Moreover, recent results obtained at
16 Cornell indicate no systemic immunological response to the
17 silica gel encapsulant (unpublished).

18 In the inventor's earliest work on cell encapsulation,

1 the single cell fungi *S. cerevisiae* was stained with
2 pyranine as a means of monitoring alcohol evolution during
3 fermentation prior to encapsulation [45,46]. In this
4 manner, we were able to optically "interface" with the
5 living cells by monitoring changes in the fluorescence
6 emission spectra. Thus, for in vivo applications, the
7 solution to both key challenges of keeping the cells alive
8 and interfacing with their metabolic functions has been
9 demonstrated.

10 Researchers at ORNL have recently demonstrated the
11 ability to attach a genetically engineered microorganism,
12 *Pseudomonas fluorescens* HK44, to a hybrid circuit and detect
13 ppb levels of naphthalene [72]. Their "critter on a chip"
14 technology, if combined with recent cell encapsulation
15 advances, could lead to the development of living biosensor
16 arrays.

17 MATECH proposed to develop and ultimately commercialize
18 a broad-based sensor platform technology to allow a wide

1 range of both chemical and biological sensing functions to
2 be performed on a single optoelectronic chip. Based upon
3 past experience in employing sol-gel derived, highly porous
4 silicate materials doped with fluorescent dyes and proteins,
5 which have already been demonstrated by both MATECH and
6 numerous other leading research groups (mostly in academia),
7 MATECH intends to integrate them into a single MEMs based
8 Optical Sensor Array. The challenges in successfully
9 accomplishing this task are enormous and the resources and
10 expertise of numerous academic and industrial collaborators
11 will be necessary. Several key disciplines need to be
12 "integrated" into the development and commercialization
13 process if it is to succeed.

14 The MEMOSA [73] technology herein proposed relies
15 heavily upon the knowledge and expertise gained in
16 developing materials for fiber-optic sensing applications.
17 Integrating numerous individual sensors into a practical and
18 cost-effective sensor system, however, requires an approach

1 that is based upon well established techniques, such as
2 integrated circuit manufacturing methods. In this regard,
3 the MEMs approach, when combined with knowledge gained from
4 fiber-optic biosensor research, is an ideal platform to
5 build complex, multifunctional devices on a single chip.

6 Referring to Fig. 6 a simple MEMs based single sensor
7 element is shown. A thin film electroluminescent light
8 source, already licensed by MATECH from OGI, is employed to
9 excite the fluorescence of dye/protein doped porous silica
10 microspheres. The emission signal is detected by a silicon
11 based photodiode which can be easily built into the silicon
12 wafer substrate. The trough can be etched into the silicon
13 wafer by well-known techniques or the walls of the trough
14 can be deposited onto the silicon wafer by well-known
15 techniques. The silicon detector element, which has an
16 inherently broad band wavelength sensitivity, can be "tuned"
17 to a specific wavelength by the deposition of an optical
18 band-pass filter on top of it. Moreover, inasmuch as a

1 single cell is square in shape, a total of three different
2 detectors (tuned to three different wavelengths) can be
3 incorporated into a single sensor element. Detection can be
4 based on the relative signal strength at each wavelength
5 selected.

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3 Referring to Fig. 7 through Fig. 12 there are
4 alternative designs such as when the individual sensor
5 elements are inverse pyramidal in shape. Once again, three
6 different detectors, tuned to three wavelengths, allows the
7 sensor to act as a crude spectrophotometer. In this design,
8 however, a thin porous mirror is applied to the top of the
9 sensor array (already demonstrated by MATECH for microsphere
10 based arrays). The high surface area doped sol-gel material
11 is deposited into each inverted pyramidal shaped element.
12 Each element can be doped with a different dye, enzyme, or
13 protein tailored to a specific species. As described in
14 sections 2 and 3, the signals from each element of the array
15 can be analyzed to produce a "map" of the unknown compound
16 or mixture and compared to an established data base of
17 references. In this manner the presence of toxic chemicals,
18 heavy metals, food born biological pathogens, biological

1 warfare agents, chemical warfare agents, and diseases known
2 to attach humans and animals can all be detected rapidly
3 from a single small sample of vapor or fluid.

4 Referring to Fig. 13 current "state of the art" biochips
5 use fluorescence consist of patterned microarrays for DNA
6 and RNA detection. These micro-arrays are, usually
7 patterned on glass slides, are inserted into large
8 analytical instruments in order to obtain detection results.

9 By integrating the entire instrument onto the chip the
10 world's smallest spectrophotometer can be created. The
11 entire instrument will be disposable. This instrumentation
12 platform can be extremely versatile inasmuch as it will be
13 portable, battery operated, and capable of deployment in
14 remote locations and even locations that are unsuitable or
15 unsafe for human presence. The key principles and
16 requirements are that: 1) all light sources must be on the
17 chip; 2) all optical detectors (at different wavelengths) be
18 on the chip; and 3) the relevant bioactive materials be on

1 the chip. The chip will only require power and will produce
2 only electrical output signals.

3 Referring to Fig. 14 the chip can potentially have two
4 modes of operation, transmission spectrophotometry and
5 fluorescence spectrophotometry. A prototype octahedral
6 "sensenel" has two different light sources, a UV
7 electroluminescent (EL) material and a white EL material and
8 six amorphous silicon detectors. Each amorphous silicon
9 detector should be tuned to a different wavelength range
10 using an optical band-pass filter coating. In transmission
11 mode, the white EL is activated and the transmission spectra
12 of the bioactive material is measured. In fluorescence mode,
13 the UV EL material is activated and the fluorescence spectra
14 of the bioactive material is measured.

15 Referring to Fig. 15 whether in transmission mode or
16 fluorescence mode, the six detectors will produce an
17 electrical output signal as a function of the light
18 intensity at each of the six detectors wavelengths that can

1 be viewed as a histogram.

2 Referring to Fig. 16 another way to plot the output
3 data is to plot it as an emission or transmission spectra
4 (depending upon the mode of operation) and curve fit the six
5 data points. Signal processing and interpretation is an
6 important aspect of the chip's design and function.

7 Referring to Fig. 17 one possible design of each sensel
8 would be inverted octagonal "pyramids" defining a depression
9 in which the bioactive material can be deposited. There are
10 numerous possible ways of depositing the bioactive material
11 to be photometrically evaluated. One method would be to use
12 microspheres. Microspheres could be placed in each well and
13 attached with adhesive. The advantage that this approach
14 offers is that the electro-optical substrate of the arrays
15 could be fabricated identically. The bioactive function of
16 each array can be customized for a specific application
17 through the selection of the microspheres to be placed in
18 it. The microspheres utilized can be porous or dense,

organic or inorganic, depending upon the specific biological and/or chemical interaction being investigated. For example, sol-gel derived porous microspheres containing a wide range of biological enzymes could be used. Alternatively, non-porous beads with fluorescent dye conjugated antibodies bound to their outer surfaces could be used for antigen detection. A very wide range of possible biological and chemical assays could be integrated into the chip.

Referring to Fig. 18 another possible design for sensels is based upon the same underlying electro-optic array, but with a significant difference--the bioactive material would be cast in place in each well. Whether using polymeric organic gels or sol-gel derived porous silica, the wet gel octagonal bioactive materials would be pipetted into each well and gelled in place. On top of the array, a thin, porous reflective polymer layer would be applied. This layer would permit analytes to permeate the bioactive gel underneath. The reflectivity of the layer would assure that

much of the light would not be lost outside the plane of the chip. The key to the BioOptix chip is the ability to have a large plurality of sensels on a single chip of very small dimensions.

Referring to Fig. 19 a simple 64 sensel chip is illustrated. As the technology advances, it should be possible to fabricate a chip of no more than 1.0 cm in size with in excess of 10,000 individual sensel elements.

Referring to Fig. 20 in conjunction with Fig. 21 in order to fully exploit the potential of the BioOptix chip, a wide range of biological assays will need to be integrated into the chip's "portfolio" of bioactive detection systems.

These include assays for molecular biology, immunoassays, enzymatic assays, and receptor-ligand assays. For molecular biology and enzymatic assays, porous sol-gel derived silica microspheres doped with the appropriate fluorophor-labeled enzymes is an attractive bioactive materials platform.

Referring to Fig. 22 it has been demonstrated using

1 microarray technology that protein-protein interactions can
2 be quantitatively measured using fluorescence. For much
3 larger detection targets, such as antigen-specific IgG,
4 surface binding interactions can be utilized using
5 microspheres.

6 The bioactive components of the BioOptix chip need to
7 be either embedded or attached to a variety of substrate
8 materials to optimize their function and assure sensitivity
9 and attain reproducible and quantifiable results.

10 One of the exciting applications of the BioOptix chip
11 might include the development of a "one drop" blood chem
12 panel-the almost instantaneous analysis of blood chemistry
13 using a single drop of blood. The market for microarray
14 technology has been growing rapidly in the past few years.
15 There are numerous companies involved in many different
16 aspects of microarray technology (see company list below).
17 Conventional microarray technology utilizes a pattern of
18 densely packed bioactive "spots" that are "spotted" onto a

1 glass slide using a robotic "spotter". After exposure to
 2 the sample that is to be analyzed, the microarray is
 3 inserted into a microarray reader, which is a large
 4 instrument with a light source and sophisticated detection
 5 system (often a CCD array). These systems are large and not
 6 portable. The BioOptix chip requires the deposition of 2
 7 different electro-luminescent light sources, six amorphous
 8 silicon photo-detectors (each with a different optical band-
 9 pass filter deposited on top of it), and 16 electrical
 10 connections to each individual sensel. All of these must be
 11 fabricated onto the surfaces of octagonal inverted pyramidal
 12 indentations. Therefore, the challenges in fabricating the
 13 electro-optic platform are considerable.

14 The output signals from each sensel will need to be
 15 processed by software capable of signal pattern recognition.
 16 Signal processing is integral to the function of the
 17 BioOptix chip.

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- 18 spectrophotometer microarray biochip platform. Current

"state of the art" biochips using fluorescence consist of patterned microarrays for DNA and RNA detection. These microarrays, usually patterned on glass slides, are inserted into large analytical instruments in order to obtain detection results. We seek to fully integrate the instrument and the biochip array into one device. By integrating the entire instrument onto the chip, we will be creating the world's smallest spectro-photometer. In addition, the entire instrument will be disposable. This instrumentation platform can be extremely versatile inasmuch as it will be portable, battery operated, and capable of deployment in remote locations and even locations that are unsuitable or unsafe for human presence.

The key principles and requirements in the design and fabrication of the BioOptix chip technology are that: 1) all light sources must be on the chip; 2) all optical detectors (at different wavelengths) must be on the chip, and; 3) the relevant bioactive materials must be on the chip. The chip

1 will only require power and will produce only electrical
2 output signals.

3 The BioOptix chip can potentially have two modes of
4 operation, transmission spectrophotometry and fluorescence
5 spectrophotometry.

6 In order to fully exploit the potential of the BioOptix
7 chip, a wide range of biological assays will need to be
8 integrated into the chip's "portfolio" of bioactive
9 detection systems. These include assays for molecular
10 biology, immunoassays, enzymatic assays, and receptor-ligand
11 assays. For molecular biology and enzymatic assays, porous
12 sol-gel derived silica microspheres doped with the
13 appropriate fluorophor-labeled enzymes is an attractive
14 bioactive materials platform. It has already been
15 demonstrated using microarray technology that protein-
16 protein interactions can be quantitatively measured using
17 fluorescence. For much larger detection targets, such as
18 antigen-specific IgG, surface binding interactions can be

1 utilized using microspheres.

2 The bioactive components of the BioOptix chip need to
3 be either embedded or attached to a variety of substrate
4 materials to optimize their function and assure sensitivity
5 and attain reproducible and quantifiable results. Materials
6 skills involved include, but are not limited to, sol-gel
7 chemistry and organic polymer chemistry.

8 Inasmuch as the BioOptix chip is essentially an array
9 to microscopic transmission and fluorescence
10 spectrophotometers, good optical engineering design and
11 performance is critical to their function.

12 Most of the assays of potential interest are
13 biochemically based. Target analytes include DNA,
14 antibodies, enzymes, and receptors. The development of,
15 and/or modification of existing assays, to the BioOptix
16 platform requirements will be extensively required.

17 One of the exciting applications of the BioOptix chip
18 might include the development of a "one drop" blood chem

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1 panel-the almost instantaneous analysis of blood chemistry
2 using a single drop of blood.

3 The market for microarray technology has been growing
4 rapidly in the past few years. There are numerous companies
5 involved in many different aspects of microarray technology
6 (see company list below). Conventional microarray technology
7 utilizes a pattern of densely packed bioactive "spots" that
8 are "spotted" onto a glass slide using a robotic "spotter".
9 After exposure to the sample that is to be analyzed, the
10 microarray is inserted into a microarray reader, which is a
11 large instrument with a light source and sophisticated
12 detection system (often a CCD array). These systems are
13 large and not portable. The BioOptix chip requires the
14 deposition of 2 different electroluminescent light sources,
15 six amorphous silicon photodetectors (each with a different
16 optical band-pass filter deposited on top of it), and 16
17 electrical connections to each individual sensel. All of
18 these must be fabricated onto the surfaces of octagonal

1 inverted pyramidal indentations. Therefore, the challenges
2 in fabricating the electro-optic platform are considerable.

3 The output signals from each sensel will need to be
4 processed by software capable of signal pattern recognition.

5 Signal processing is integral to the function of the
6 BioOptix chip has been described.

7 While this invention has been particularly shown and
8 described with references to preferred embodiments thereof,
9 it will be understood by those skilled in the art that
10 various changes in form and details may be made therein
11 without departing from the spirit and scope of the invention
12 as defined by the appended claims

13 It should be noted that the sketches are not drawn to
14 scale and that distance of and between the figures are not
15 to be considered significant.

16 Accordingly it is intended that the foregoing
17 disclosure and showing made in the drawing shall be
18 considered only as an illustration of the principle of the

1 present invention.

2